

Total Synthesis of (\pm)-Scopadulcic Acid A. An Illustration of the Utility of Palladium-Catalyzed Polyene Cyclizations

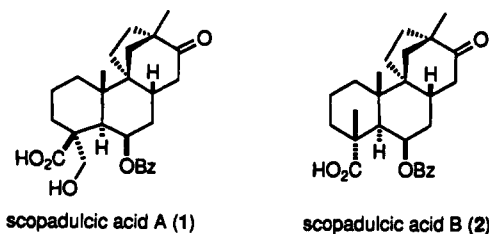
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Summary: The first total synthesis of (\pm)-scopadulcic acid A has been accomplished using an intramolecular bis-Heck cyclization to form the B-D rings of this tetracyclic diterpene with complete stereocontrol.

The plant *Scoparia dulcis* has long been used in the traditional practice of medicine in Paraguay, India, and Taiwan.² In a search for biologically active substances from the Paraguayan crude drug "Typychá kuratú", Hayashi and co-workers described in 1987 the isolation from *Scoparia dulcis* L. (Scrophulariaceae) of two structurally novel tetracyclic diterpene acids, the scopadulcic acids A (1) and B (2).³ Preliminary investigations reveal a remarkable pharmacological spectrum for the scopadulcic diterpenes: in vitro and in vivo antiviral and antitumor activities and powerful inhibitory activity against H⁺,K⁺-adenosine triphosphatase (the proton pump for gastric acid secretion).^{3c,4} We recently reported the total synthesis of (\pm)-scopadulcic acid B, which was the first total synthesis of a scopadulcic diterpene.⁵ Herein we record the first total synthesis of scopadulcic acid A by a strategy that is more direct than our first generation entry to this unusual class of pharmacological agents.



A bis-Heck cyclization of a methylenecycloheptenyl aryl ketone was the central step in our synthesis of scopadulcic

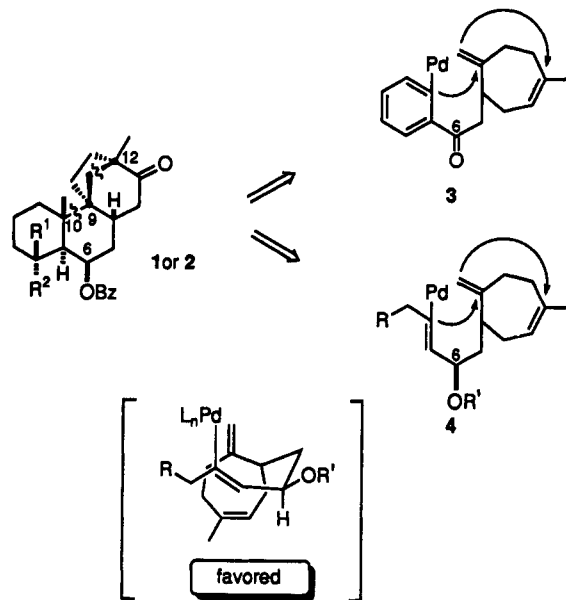


Figure 1. Synthesis plan.

acid B (Figure 1, 3 \rightarrow 2).^{5,6} Since an aromatic ring is not an ideal structural template for evolving the two quaternary centers of the scopadulcic A ring, our second generation approach adopted a seco-A ring strategy in which bis-cyclization of a monocyclic trienyl iodide would be the central step (Figure 1, 4 \rightarrow 1 or 2). An additional key objective of this second generation strategy was to control face selection in the initial insertion of the exomethylene group by incorporating a β -alkoxy substituent at C(6) in the cyclization substrate 4. The rationale for this choice is depicted in Figure 1, where a preferred eclipsed orientation⁷ of the C-Pd σ -bond and the exomethylene group places the alkoxy substituent in an unhindered equatorial environment.

Methylenecycloheptene aldehyde 9, the precursor of the cyclization substrate 12, was assembled using a divinylcyclopropane rearrangement strategy similar to that employed in our earlier synthesis of scopadulcic acid B.^{5,8} One change merits note. We found it considerably more convenient not to resolve the stereoisomers of cyclopropyl bromide 5, but rather to use the readily available isomer mixture⁹ to prepare the cyclopropyl ketones 6. Thermal

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(1) Current Address: Bristol Myers Squibb, One Squibb Drive, P. O. Box 191, New Brunswick, NJ 08903.

(2) Gonzales Torres, D. M. *Catalogo de Plantas Medicinales (y Alimenticias y Utiles) Usada en Paraguay*, Asunción: Paraguay, 1986; p 394. Chow, S. Y.; Chen, S. M.; Yang, C. M.; Hsu, H. *J. Formosan Med. Assoc.* 1974, 73, 729. Satyanarayana, K. *J. Ind. Chem. Soc.* 1969, 46, 765.

(3) (a) Hayashi, T.; Kishi, M.; Kawasaki, M.; Arisawa, M.; Shimizu, M.; Suzuki, S.; Yoshizaki, M.; Morita, N.; Tezuka, Y.; Kikuchi, T.; Berganza, L. H.; Ferro, E.; Basualdo, I. *Tetrahedron Lett.* 1987, 28, 3693. (b) Hayashi, T.; Kishi, M.; Kawasaki, M.; Arisawa, M.; Morita, N. *J. Nat. Prod.* 1988, 51, 360. (c) Hayashi, T.; Asano, S.; Mizutani, M.; Takeguchi, N.; Kojima, T.; Okamura, K.; Morita, N. *J. Nat. Prod.* 1991, 54, 802.

(4) (a) Hayashi, K.; Niwayama, S.; Hayashi, T.; Nago, R.; Ochiai, H. and Morita, N. *Antiviral Res.* 1988, 9, 345. (b) Asano, S.; Mizutani, M.; Hayashi, T.; Morita, N.; Takeguchi, N. *J. Biol. Chem.* 1990, 265, 22167. (c) Hayashi, T.; Okamura, K.; Kakemi, M.; Asano, S.; Mizutani, M.; Takeguchi, N.; Kawasaki, M.; Tezuka, Y.; Kikuchi, T.; Morita, N. *Chem. Pharm. Bull.* 1990, 38, 2740. (d) Hayashi, T.; Hayashi, K.; Uchida, K.; Niwayama, S.; Morita, N. *Chem. Pharm. Bull.* 1990, 38, 239. (e) Hayashi, T.; Asano, S.; Mizutani, M.; Takeguchi, N.; Kojima, T.; Okamura, K. and Morita, N. *J. Nat. Prod.* 1991, 54, 802. (f) Hayashi, K.; Hayashi, T.; Morita, N. *Phytotherapy Res.* 1992, 6, 6.

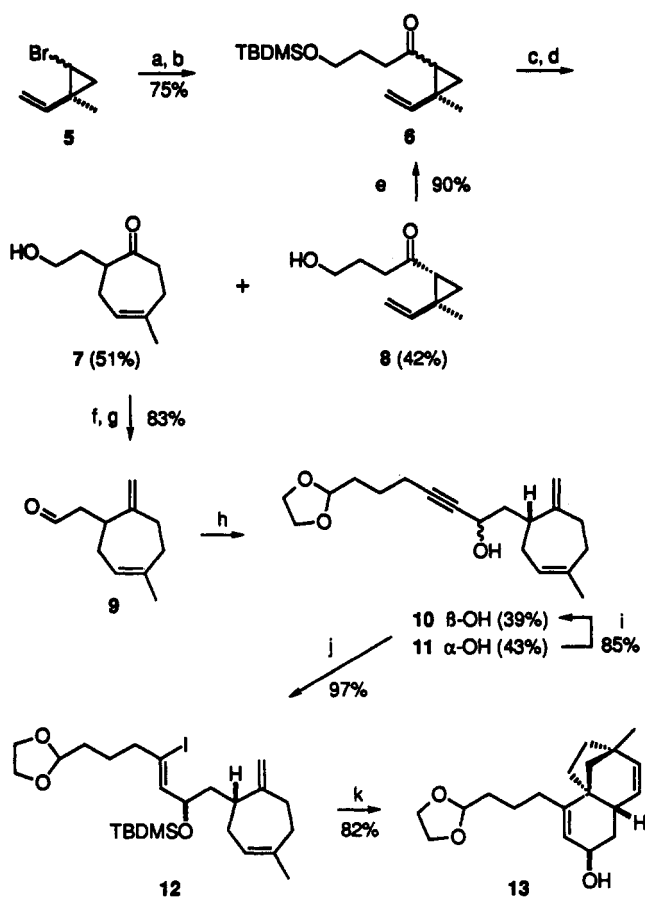
(5) Overman, L. E.; Ricca, D. J.; Tran, V. D. *J. Am. Chem. Soc.* 1993, 115, 2042-44.

(6) For a recent brief review of our work in the area of palladium-catalyzed polyene cyclizations, see: Overman, L. E.; Abelman, M. M.; Kucera, D. J.; Tran, V. D.; Ricca, D. *J. Pure Appl. Chem.* 1992, 64, 1813.

(7) Abelman, M. M., Overman, L. E.; Tran, V. D. *J. Am. Chem. Soc.* 1990, 112, 6959.

(8) (a) Piers, E.; Nagakura, I. *Tetrahedron Lett.* 1976, 3247. (b) Marino, J. P.; Browne, L. *J. Tetrahedron Lett.* 1976, 3245. (c) Wender, P. A.; Filosa, M. P. *J. Org. Chem.* 1976, 41, 3490. (d) Wender, P. A.; Eissenstat, M. A.; Filosa, M. P. *J. Am. Chem. Soc.* 1979, 101, 2196. (e) Piers, E.; Burmeister, M. S.; Ressig, H.-U. *Can. J. Chem.* 1986, 64, 180. (f) Piers, E. In *Comprehensive Organic Synthesis*; Trost, B. M., Fleming, I., Eds.; Pergamon: Oxford, 1991; Vol. 5, Chapter 8.2.

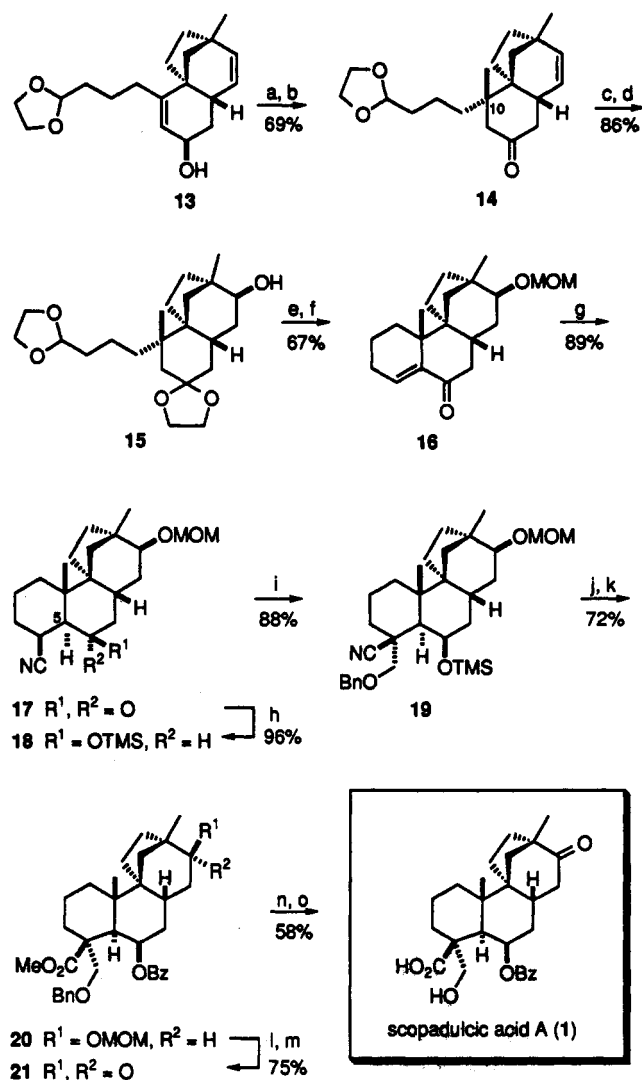
(9) (a) Skatellbol, L. *J. Org. Chem.* 1964, 29, 2951. (b) Seyferth, D.; Yamazaki, H.; Alleston, D. *J. Org. Chem.* 1963, 28, 703.

Scheme I^a

^a Reaction conditions: (a) *t*-BuLi, Et₂O, -78 °C; TBDMSO(CH₂)₃CHO, -78 → 23 °C; (b) Swern oxidation; (c) TMSOTf, Et₃N, CH₂Cl₂, 0 °C; (d) PhH, reflux; PPTS, *i*-PrOH-H₂O, 23 °C; TBAF, THF, 23 °C; (e) NaOMe, MeOH, reflux (*Z*:*E* = 3:4); TBSCl, imidazole; (f) Ph₃P=CH₂, DMSO, 23 °C, 91%; (g) Swern oxidation, 91%; (h) (OCH₂CH₂O)CH(CH₂)₃C=Cl, THF, -78 → 23 °C; separation on silica gel; (i) Ph₃P, DEAD, PhCO₂H, -20 → 23 °C; K₂CO₃, MeOH; (j) Red-Al, THF, 23 °C; NIS, -70 → 23 °C; TBDMSCl, imidazole, DMF; (k) 10% Pd(OAc)₂-Ph₃P (1:2), Ag₂CO₃, THF, reflux; TBAF, THF, 23 °C.

rearrangement of the enoxysilanes derived from these intermediates afforded, after desilylation, a readily separable mixture of cycloheptenone 7 and the *trans*-cyclopropyl ketone 8.¹⁰ Recycling this latter intermediate allowed 9 to be prepared in 75% overall yield from the ~1:1 mixture of cyclopropyl ketones 6.

Reaction of aldehyde 9 with [6-(ethylenedioxy)-1-hexynyl]lithium¹¹ provided a ~1:1 mixture of propargylic alcohols 10 and 11, which could be separated by careful chromatography on silica gel. Mitsunobu inversion¹² of the undesired α -epimer 11 allowed 10 to be prepared in 78% yield from 9. Reduction of 10 with sodium bis(2-methoxyethoxy)aluminum hydride (Red-Al),¹³ iodination of the resulting vinylalanate, and protection of the allylic alcohol gave the *Z* trienyl iodide cyclization substrate 12. The critical bis-Heck cyclization of 12 proceeded cleanly in refluxing THF in the presence of 10% Pd(OAc)₂, 20%

Scheme II^a

^a Reaction conditions: (a) TPAP, NMO, CH₂Cl₂-CH₃CN, 23 °C, 88%; (b) Me₂Zn, LiBr, Ni(acac)₂, Et₂O, 23 °C; 78%; (c) (CH₂OH)₂, (OCH₂CH₂O)CHOMe Amberlyst 15, Me₃CN; 23 °C, 91%; (d) *m*-CPBA, NaHCO₃, CH₂Cl₂, 0 °C; LiAlH₄, THF, 23 °C, 94%; (e) 20% aqueous HCl-MeOH, 65 °C, 87%; (f) MOMCl, (*i*-Pr)₂NEt, CH₂Cl₂, 77%; (g) Et₂AlCN, THF, 0 °C; TMSCl, Et₃N, 0 °C; dilute aqueous HCl, 23 °C; (h) LiAlH₄, THF, -78 °C, 99%; TMSCl, DMAP, pyridine-CH₂Cl₂, 0 °C, 97%; (i) LDA, THF, 0 °C; BnOCH₂Br, -78 → 0 °C; (j) KOH, K₂CO₃, 140 °C; CH₂N₂, Et₂O, 0 → 23 °C, 82%; (k) BzCl, DMAP, pyridine, 100 °C; (l) HCl, MeOH, 70 °C, 82%; (m) PCC, 4-Å molecular sieves; CH₂Cl₂, 23 °C, 91%; (n) *n*-PrSLi, DMPU, 23 °C, 60%; (o) H₂, Pd-C, EtOAc-MeOH, 23 °C, ~100%.

Ph₃P, and 2 equiv of Ag₂CO₃ to give, after desilylation, a *single* tricyclic product 13 in 82% yield. That the initial insertion occurred with complete stereochemical fidelity in the desired sense was established in a related series (OTBDMS instead of ethylene acetal) by X-ray analysis.¹⁴ The efficient sequence summarized in Scheme I provided 13 on multigram scales in 31% overall yield from the mixture of cyclopropyl bromides 5 (16% overall from isoprene).

The efficient elaboration of 13 to the scopadulan ring system is summarized in Scheme II. The critical angular methyl group at C(10) was introduced by oxidation of 13 to form the corresponding enone¹⁵ followed by reaction of

(10) All intermediates were fully characterized by ¹H and ¹³C NMR, IR, and MS analysis. The elemental composition of analytical samples of new compounds was confirmed by combustion analysis or high-resolution mass spectrometry. Yields refer to isolated, purified products unless noted otherwise.

(11) Prepared from 5-hexynol by oxidation (ClCOCOCl, Me₂SO, Et₃N) followed by acetalization (HOCH₂CH₂OH, *p*-TsOH).

(12) Hughes, D. L. *Org. React. (N.Y.)* 1992, 42, Chapter 2.

(13) Denmark, S. E.; Jones T. K. *J. Org. Chem.* 1982, 47, 4595.

(14) Kucera, D. J. Ph.D. Dissertation, University of California, Irvine, 1991.

(15) Griffith, W. P.; Ley, S. V. *Aldrichim. Acta* 1990, 23, 13.

this intermediate with $\text{Me}_2\text{Zn}/\text{Ni}(\text{acac})$.¹⁶ Conjugate addition of methyl occurred with complete facial selectivity from the same face as the one-carbon bridge to give 14 in 69% overall yield from 13.¹⁷ The oxygen functionality of the C ring was next developed by protection of the C(6) ketone of 14, epoxidation of the resulting bis-ketal, and regioselective reduction of the derived β -epoxide to provide 15. Treatment of this intermediate with dilute methanolic HCl afforded the crystalline (mp 115–117 °C) tetracyclic enone in one step. Standard protection of the hydroxy group then provided the MOM ether 16 in 67% yield. Enone 16 is a potentially versatile intermediate for preparing a variety of scopadulan diterpenes and analogs and is available on preparative scales in 15 total steps and 5% overall yield from 5.

(\pm)-Scopadulcic acid A was prepared from the tetracyclic enone 16 as follows. Reaction of 16 with Et_2AlCN and Me_3SiCl , followed by hydrolysis with dilute HCl, provided 17 (mp 171–172 °C) in 89% yield.¹⁸ Reduction of 17 with LiAlH_4 at –78 °C provided exclusively the axial C(6) alcohol, which was silylated to provide 18. The remaining carbon appendage at C(4) was then introduced, with complete stereocontrol, by reaction of the lithium salt of nitrile 18 with (benzyloxy)methyl bromide to afford 19 in 84% yield from 17. Although the hindered axial nitrile group of 19 was stable to many common hydrolysis conditions, it could be transformed to the carboxylate salt by heating a mixture of 19, KOH, and K_2CO_3 at 140 °C. Careful acidification, esterification of the resulting acid, and benzylation of the liberated C(6) alcohol group with

BzCl and DMAP at 100 °C in pyridine provided 20 in 72% overall yield. Using conventional transformations, tetracycle 20 was finally converted to the natural diterpene acid by transformation of the MOM ether to a carbonyl group, followed by sequential deprotection of the carboxylic acid and alcohol functionalities of 21. Although these final stages are yet to be fully optimized, (\pm)-scopadulcic acid A (1), mp 237–239 °C dec, was available from 20 in ~45% overall yield. Synthetic 1 showed 500 MHz ^1H NMR, 125 MHz ^{13}C NMR, and chromatographic properties that were indistinguishable from those of an authentic sample of 1.

This more direct total synthesis route to the scopadulan diterpenes successfully addresses several of the shortcomings of our first generation approach. Most notable is the complete stereocontrol now realized in the critical bis-Heck cyclization and the more direct elaboration from an acyclic precursor of the two quaternary centers of the scopadulan A ring. These improvements will facilitate the synthesis of congeners of 1 and 2 for future pharmacological investigation.

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Supplementary Material Available: Characterization data for new compounds and ^1H and ^{13}C NMR spectra of natural and synthetic scopadulcic acid A (9 pages). This material is contained in libraries on microfiche, immediately follows this article in the microfilm version of the journal, and can be ordered from the ACS; see any current masthead page for ordering information.

(16) Luche, J.-L.; Petrier, C.; Lansard, J.-P.; Greene, A. E. *J. Org. Chem.* 1983, 48, 3837.

(17) In contrast to a related tetracyclic enone intermediate in our earlier scopadulcic acid B synthesis,⁵ conjugate addition of methyl nucleophiles to the trisubstituted enone derived from 13 was readily accomplished, e.g., also with LiMe_2Cu .

(18) Nagata, W. *Org. React. (N.Y.)* 1977, 25, 255.